

# InertSustainSwift C8

## Technical Data

## Physical Properties

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●Silica	: ES (Evolved Surface) Silica Gel
●Particle Size	: 1.9 μm, 3 μm, 5 μm
●Surface Area	: 200 m <sup>2</sup> /g
●Pore Size	: 200 Å (20 nm)
●Pore Volume	: 1.00 mL/g
●Bonded Phase	: Octyl Groups
●End-capping	: Complete
●Carbon Loading	: 6 %
●pH Range	: 1 ~ 10
●USP Code	: L7

# Benefits of InertSustainSwift C8

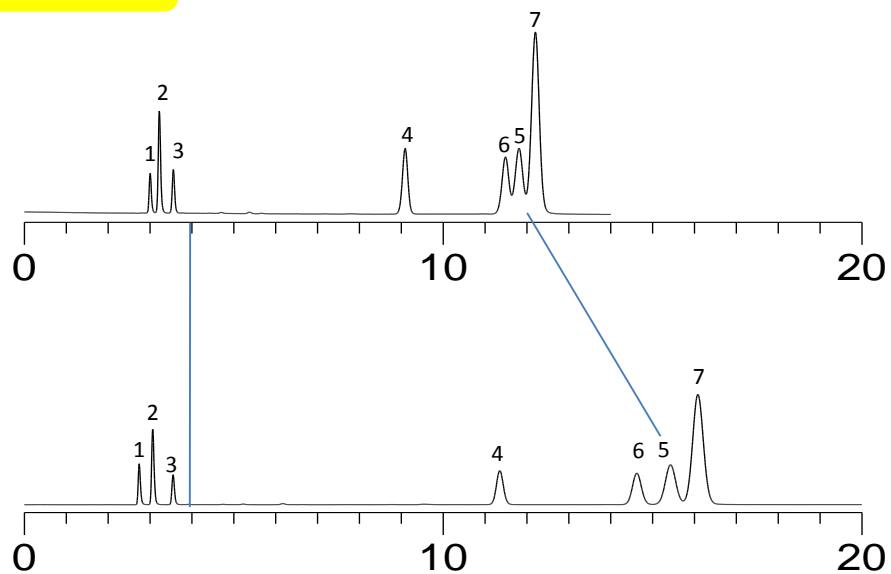
- **Highly inert packing material results in less tailing of peaks for virtually any type of analytes**
- **Extreme resistance to low and high pH mobile phases**
- **Rapid elution of samples in isocratic methods**
- **Endlessly reproducible from column-to-column and batch-to-batch**

# Comparison of Retentivity Between InertSustainSwift C8 & InertSustain C8

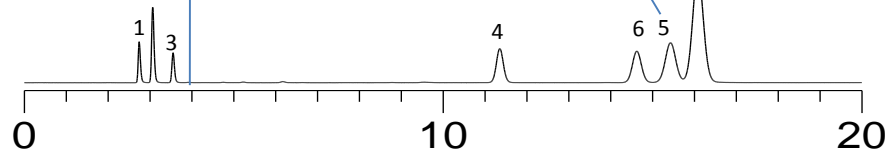
The retentivity of InertSustainSwift C8 is weaker than InertSustain C8 for neutral and especially for hydrophobic compounds. The weak retentivity is generated due to the difference in physical properties between InertSustainSwift C8 (surface area: 200 m<sup>2</sup>/g, carbon load: 6 %) and InertSustain C8 (surface area: 350 m<sup>2</sup>/g, carbon load: 8 %).

## Highly Hydrophilic – Neutral Compounds

### InertSustainSwift C8



### InertSustain C8

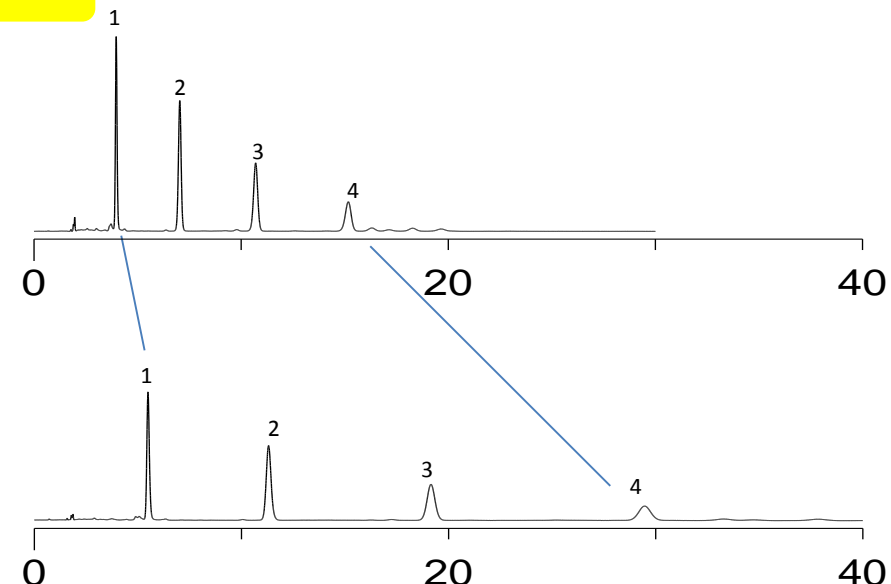


#### Conditions

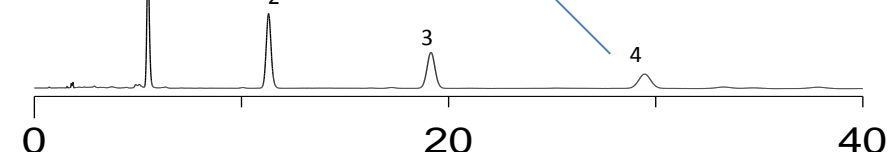
Column size	: 5 μm, 250 x 4.6 mm I.D.	1. Uracil
Eluent	: A) CH <sub>3</sub> OH	2. Caffeine
	B) H <sub>2</sub> O	3. Phenol
	A / B = 80/20, v / v	4. n-Butylbenzene
Flow Rate	: 1.0 mL/min	5. n-Amylbenzene
Col. Temp.	: 40 °C	6. o-Terphenyl
Detection	: UV 254 nm	7. Triphenylene

## Hydrophobic Compounds

### InertSustainSwift C8



### InertSustain C8



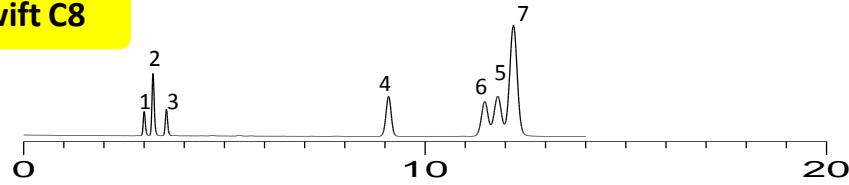
#### Conditions

Column size	: 5 μm, 150 x 4.6 mm I.D.	1. Vitamin A
Eluent	: A) CH <sub>3</sub> CN	2. Vitamin D3
	B) H <sub>2</sub> O	3. Vitamin E(α)
	A / B = 90/10, v / v	4. Vitamin K1
Flow Rate	: 1.0 mL/min	
Col. Temp.	: 40 °C	
Detection	: UV 300 nm	

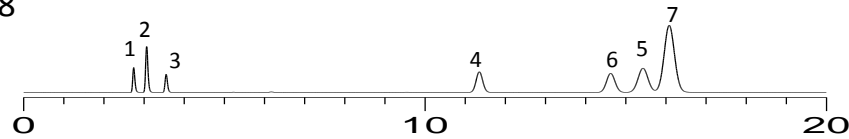
# Comparison of Retentivity Between InertSustainSwift C8 & Other C8 columns

InertSustainSwift C8 offer moderate retentivity while comparing with GL Sciences' C8 columns.

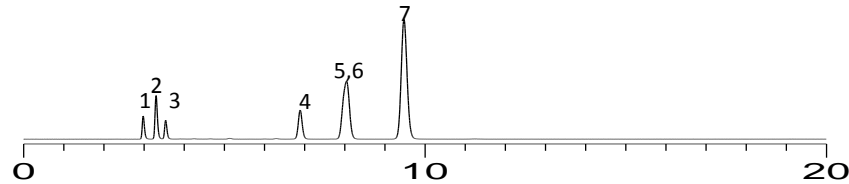
## InertSustainSwift C8



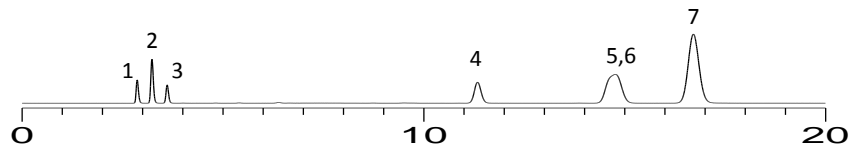
## InertSustain C8



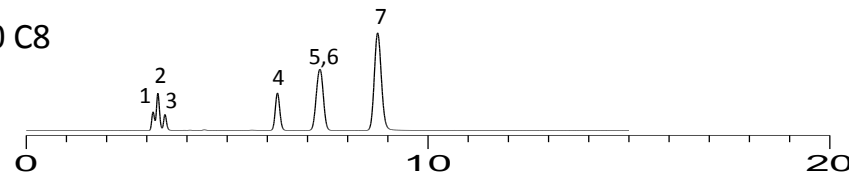
## Inertsil C8-4



## Inertsil C8-3



## Inertsil WP300 C8



### Conditions

Column size : 5  $\mu$ m, 250 x 4.6 mm I.D.

Eluent : A) CH<sub>3</sub>OH

B) H<sub>2</sub>O

A / B = 80/20, v / v

Flow Rate : 1.0 mL/min

Col. Temp. : 40 °C

Detection : UV 254 nm

1. Uracil

2. Caffeine

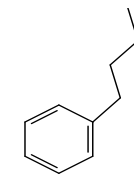
3. Phenol

4. n-Butylbenzene

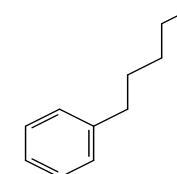
5. n-Amylbenzene

6. o-Terphenyl

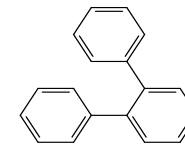
7. Triphenylene



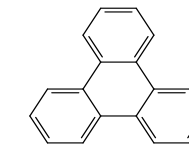
4. n-Butylbenzene



5. n-Amylbenzene



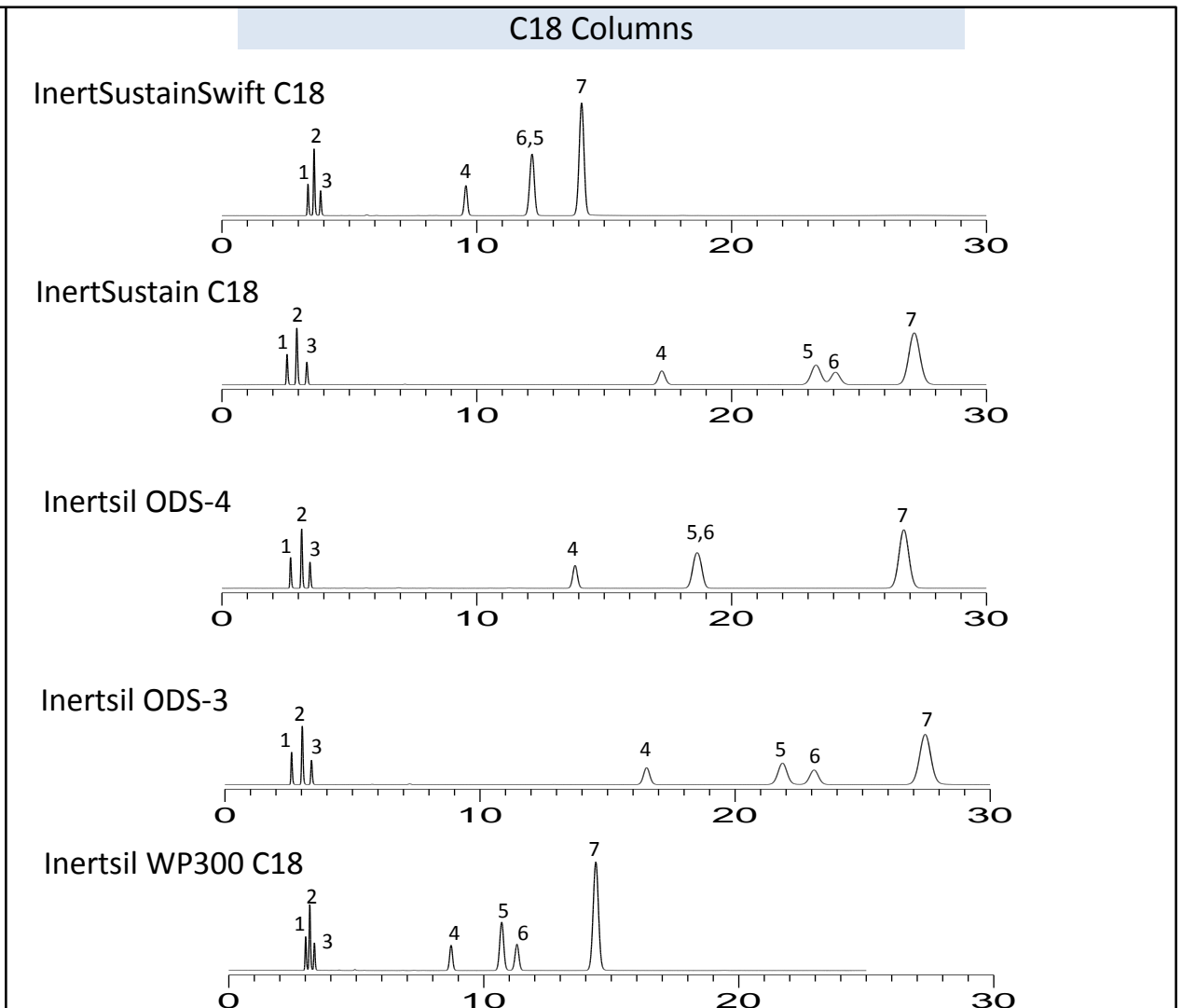
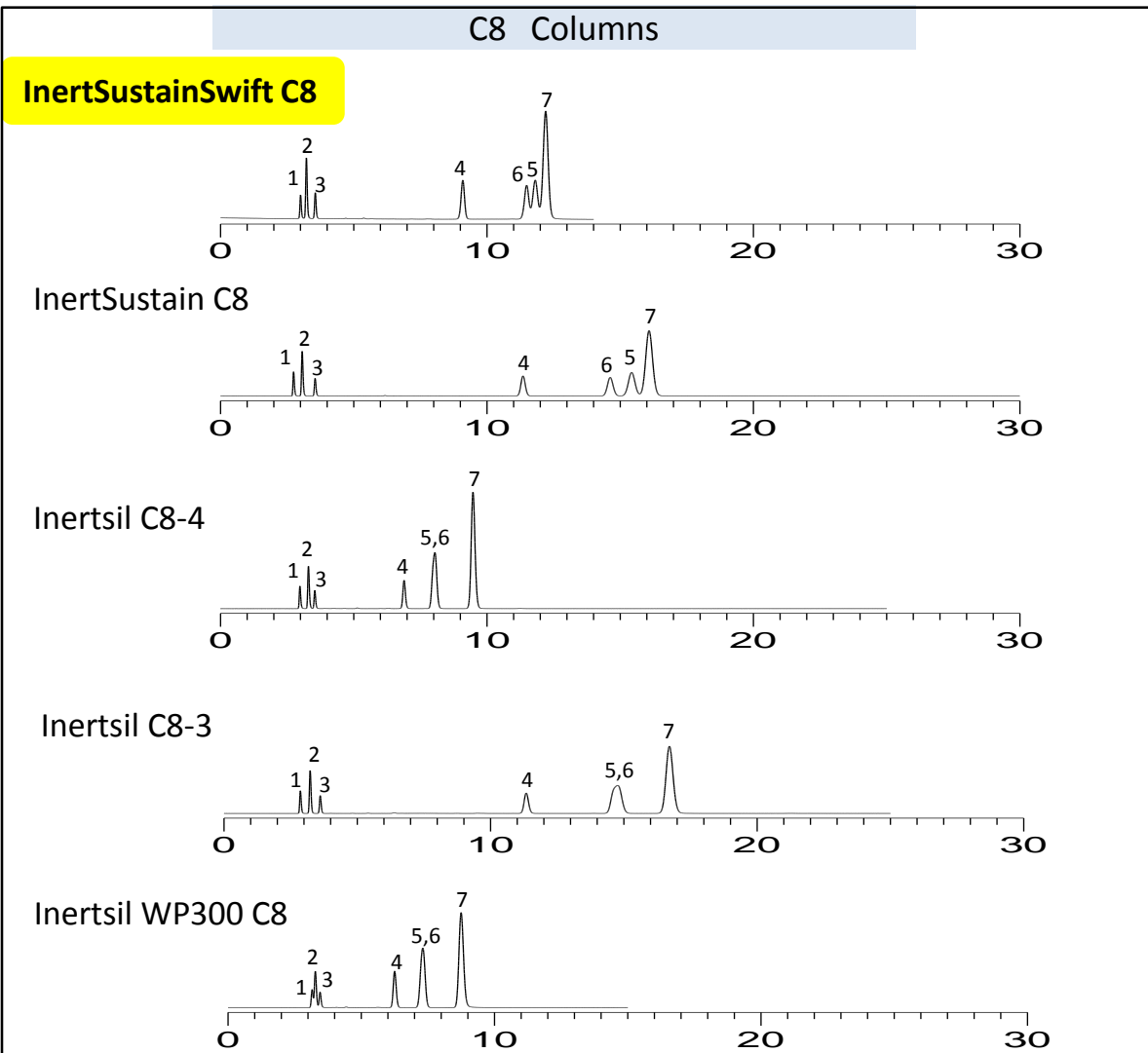
6. o-Terphenyl



7. Triphenylene

# Comparison of Retentivity Between InertSustainSwift C8 & Other C18 columns

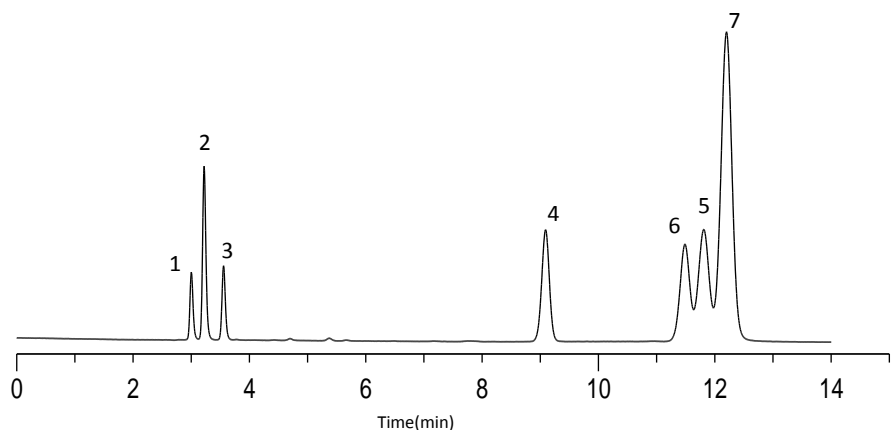
This example shows a difference in selectivity and retentivity between C18 and C8 columns which were tested under the same analytical conditions.



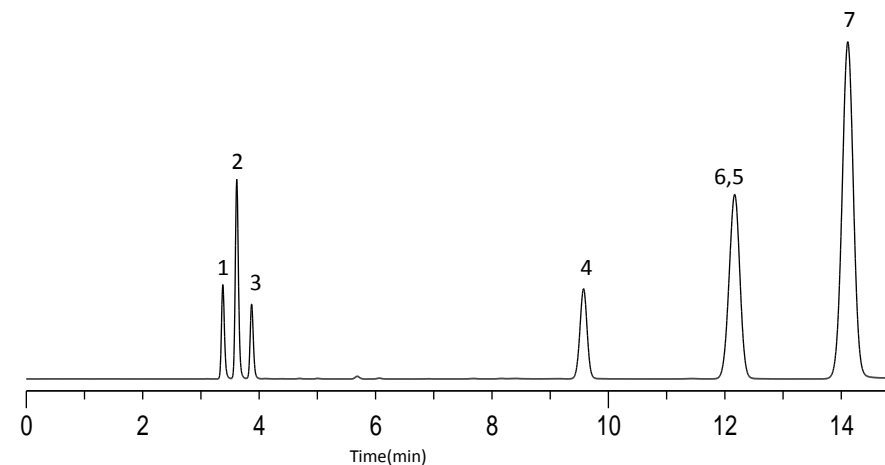
# Comparison Between InertSustainSwift C18 & C8

Under this analytical conditions, the retentivity is similar between both columns while the InertSustainSwift C8 provided better separation, which can be used to optimize selectivity or analysis time.

**InertSustainSwift C8**  
Carbon Load: 6.0 %



**InertSustainSwift C18**  
Carbon Load: 9.0 %



## Conditions

Column size : 5  $\mu$ m, 250 x 4.6 mm I.D.  
Eluent : A) CH<sub>3</sub>OH  
B) H<sub>2</sub>O  
A/B = 80/20, v/v  
Flow Rate : 1.0 mL/min  
Col. Temp. : 40 °C  
Detection : UV 254 nm

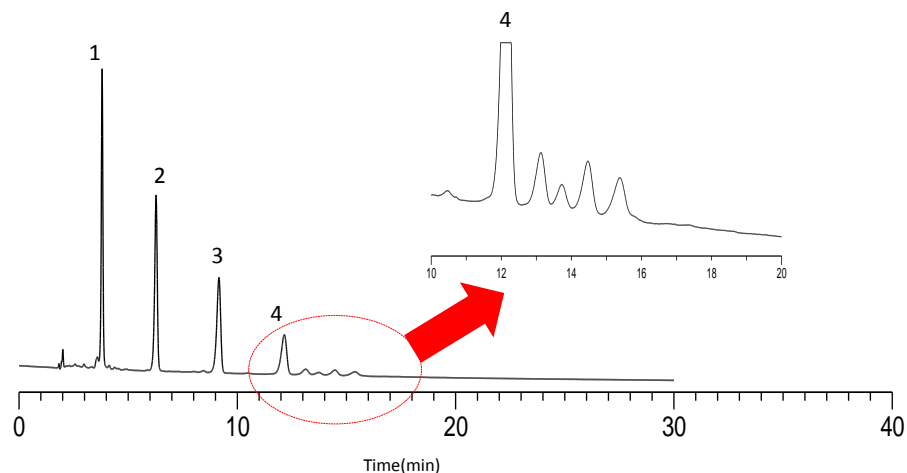
Sample:  
1.Uracil  
2.Caffeine  
3.Phenol  
4.n-Butylbenzene  
5.n-Amylbenze  
6.o-Terphenyl  
7.Triphenylene

- Sample No.4, n-Butylbenzene and Sample No.5, n-Amylbenzene were used to determine the hydrophobic property of the column. n-Amylbenzene elutes later against n-Butylbenzene when the hydrophobicity of the column is high.
- Stereoselectivity is indicated by Sample No.6, o-Terphenyl and Sample No.7, Triphenylene. O-Terphenyl has a twisted tertiary structure and Triphenylene has a planar structure. Triphenylene elutes later against o-Terphenyl when the stereoselectivity of the column is high.

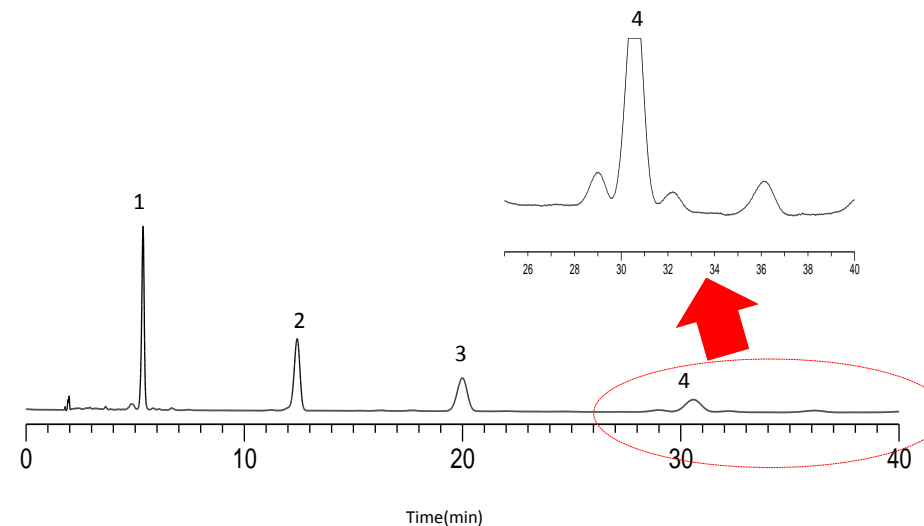
# Comparison Between InertSustainSwift C18 & C8

Many chromatographers prefer a C8 column when an ODS phase shows excessive retention values. In this example, the InertSustainSwift C8 provides shorter analysis time with better selectivity on the impurities eluted after sample no. 4 due to the shorter alkyl chain length of the stationary phase.

**InertSustainSwift C8**  
**Carbon Load: 6.0 %**



**InertSustainSwift C18**  
**Carbon Load: 9.0 %**



**Conditions**

Column size : 5  $\mu$ m, 150 x 4.6 mm I.D.  
Eluent : A) CH<sub>3</sub>CN  
          B) H<sub>2</sub>O  
          A/B = 90/10, v/v  
Flow Rate : 1.0 mL/min  
Col. Temp. : 40 °C  
Detection : UV 300 nm

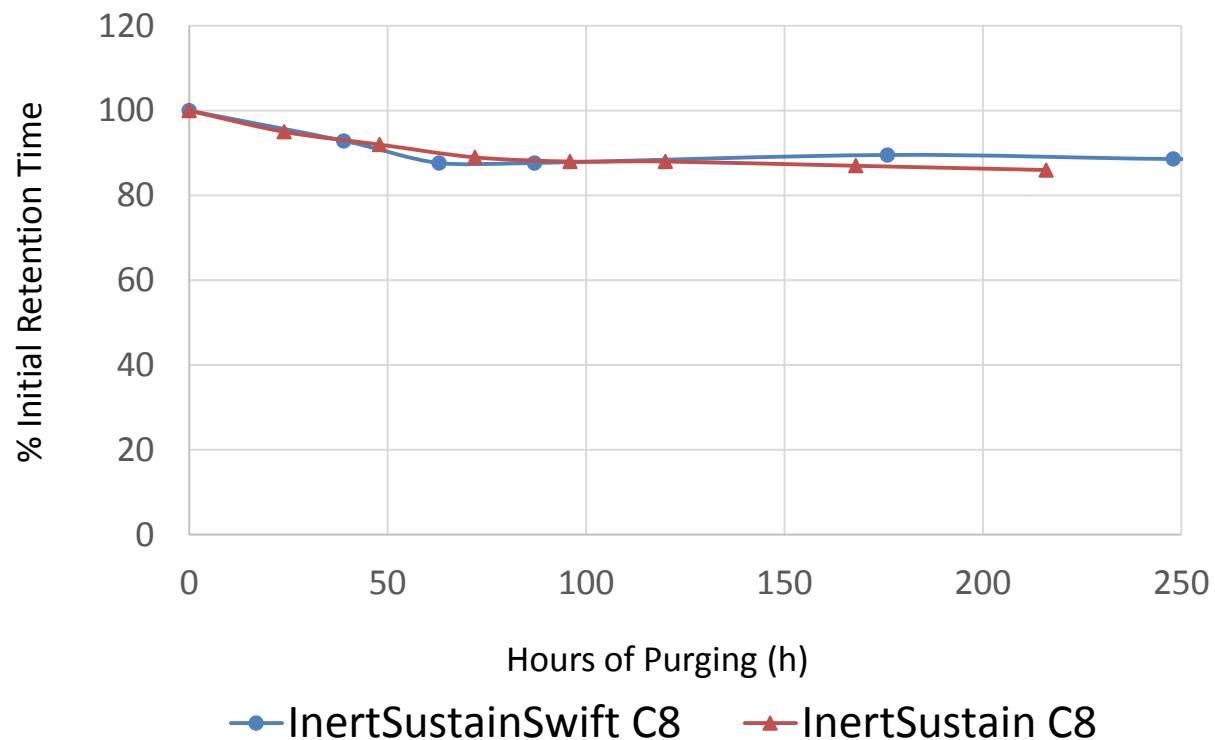
1. Vitamin A  
2. Vitamin D3  
3. Vitamin E( $\alpha$ )  
4. Vitamin K1



# Durability Testing at Low pH

InertSustainSwift C8 columns are stable over a wide pH range, just like InertSustain C8.

## Low pH Resistance Test



### Conditions

Purging Solvent : 0.1 % TFA (pH 2.0)/CH<sub>3</sub>CN = 90/10, v/v  
Column Temp. : 60 °C  
Sample : Naphthalene

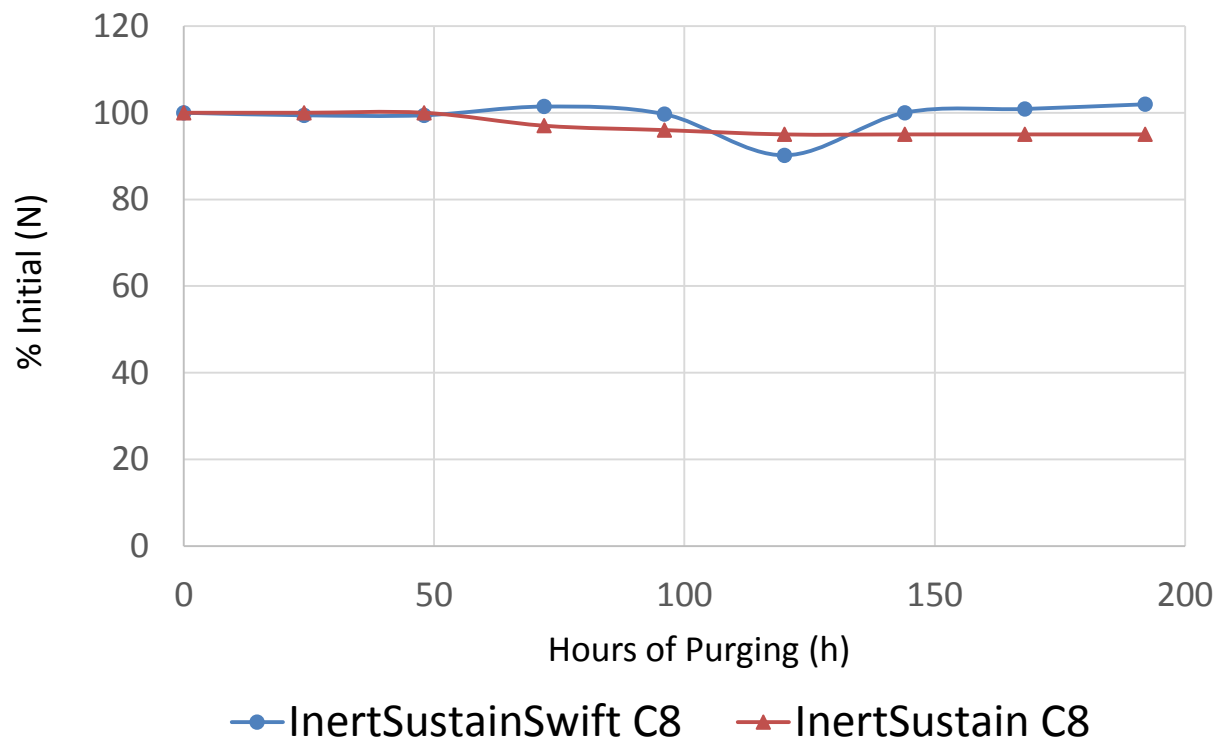
### Testing Procedure

- 1) Purging solvent is introduced into column.
- 2) The column is then flushed with 10 % CH<sub>3</sub>CN .
- 3) Naphthalene is used to verify the % initial retention time.

# Durability Testing at High pH

InertSustainSwift C8 columns are stable over a wide pH range, just like InertSustainSwift C18.

## High pH Resistance Test



### Conditions

Purging Solvent : 50 mM TEA (pH 9.0)/CH<sub>3</sub>OH = 70/30, v/v  
Column Temp. : 50 °C  
Sample : Naphthalene

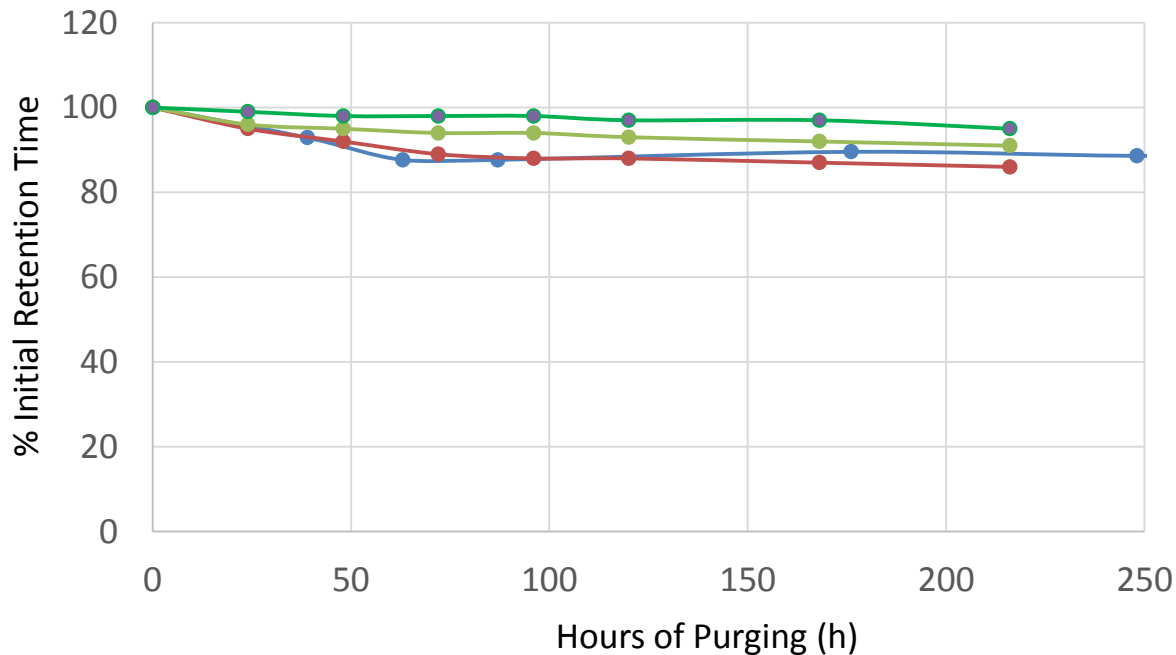
### Testing Procedure

- 1) Purging solvent is introduced into column.
- 2) The column is then flushed with 30 % CH<sub>3</sub>OH .
- 3) Naphthalene is used to verify the efficiency (N).

# Comparison of Durability Between InertSustainSwift C8 & Other C8 columns

InertSustainSwift C8 columns are stable over a wide pH range, just like InertSustain C8.

## Low pH Resistance Test



### Conditions

Purging Solvent : 0.1 % TFA (pH 2.0)/CH<sub>3</sub>CN = 90/10, v/v  
Column Temp. : 60 °C  
Sample : Naphthalene

### Testing Procedure

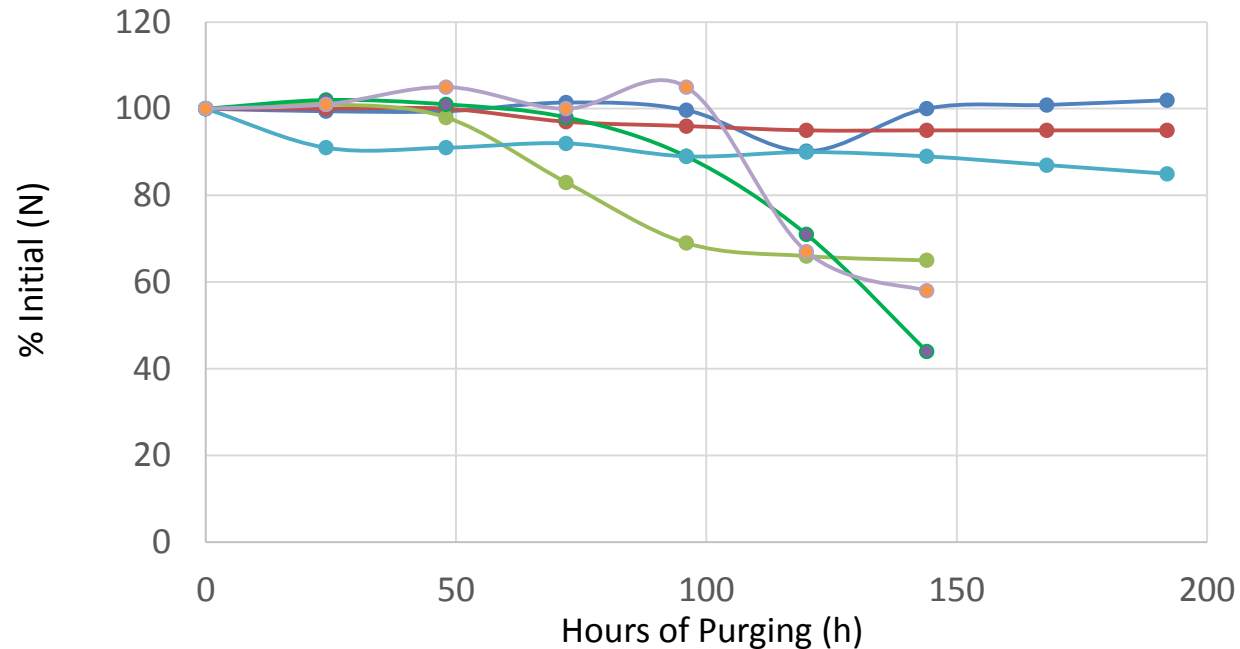
- 1) Purging solvent is introduced into column.
- 2) The column is then flushed with 10 % CH<sub>3</sub>CN .
- 3) Naphthalene is used to verify the % initial retention time.

—●— InertSustainSwift C8      —●— InertSustain C8  
—●— Inertsil C8-4            —●— Inertsil C8-3

# Comparison of Durability Between InertSustainSwift C8 & Other C8 columns

InertSustainSwift C8 columns are stable over a wide pH range, just like InertSustain C8.

## High pH Resistance Test



### Conditions

Purging Solvent : 50 mM TEA (pH 9.0)/CH<sub>3</sub>OH = 70/30, v/v  
Column Temp. : 50 °C  
Sample : Naphthalene

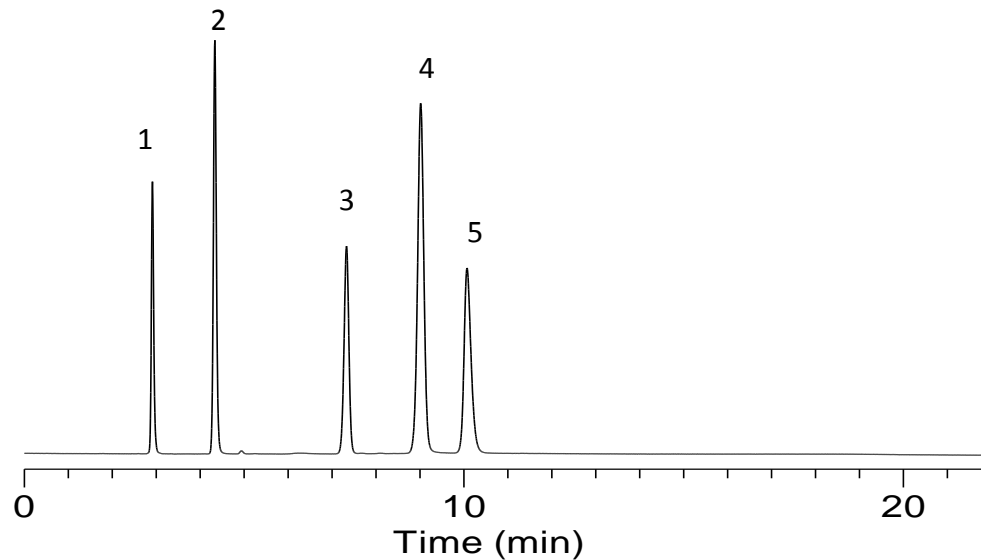
### Testing Procedure

- 1) Purging solvent is introduced into column.
- 2) The column is then flushed with 30 % CH<sub>3</sub>OH .
- 3) Naphthalene is used to verify the efficiency (N).

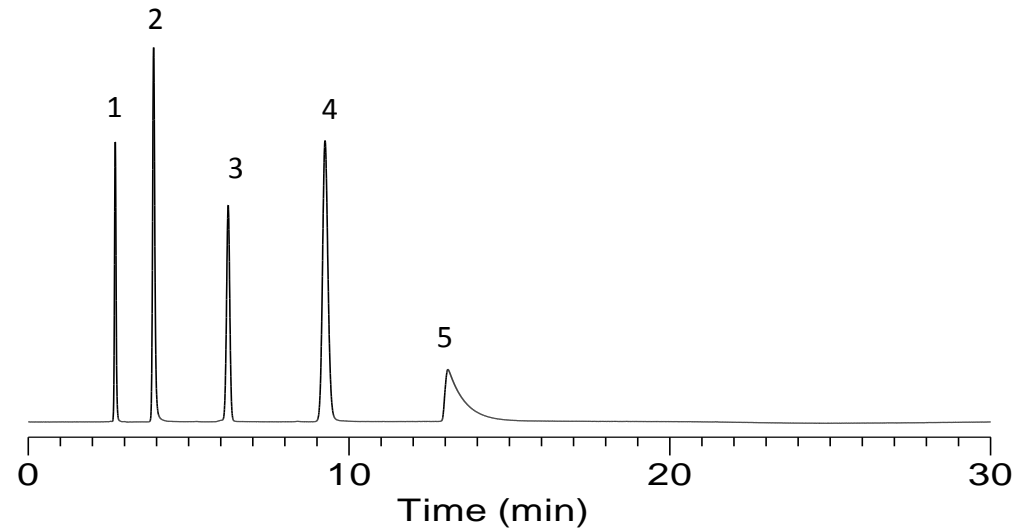
# Comparison of Strong Basic Compound Test

Dextromethorphan is a strong basic compound. Severe tailing can be confirmed when the packing material contains residual silanol groups.

## InertSustainSwift C8



## Hypersil BDS C8



### Conditions

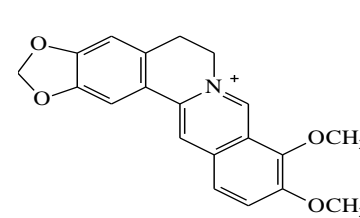
Eluent : A) CH<sub>3</sub>CN  
B) 25mM K<sub>2</sub>HPO<sub>4</sub> (pH 7.0, KH<sub>2</sub>PO<sub>4</sub>)  
A/B = 30/70, v/ v

Flow Rate : 1.0 mL/min

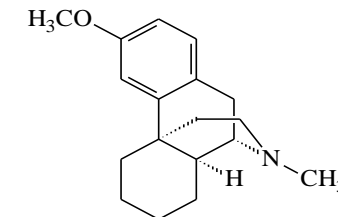
Col. Temp. : 40 °C

Detection : UV 230 nm

Sample : 1.Uracil 2.Pyridine 3.Phenol  
4.Berberine chloride 5.Dextromethorphan



4. Berberine chloride

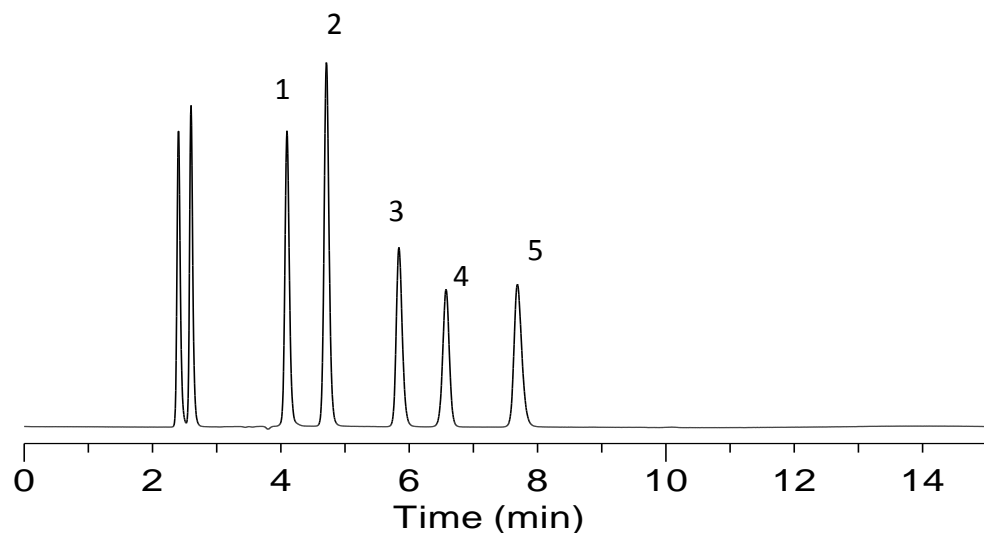


5. Dextromethorphan

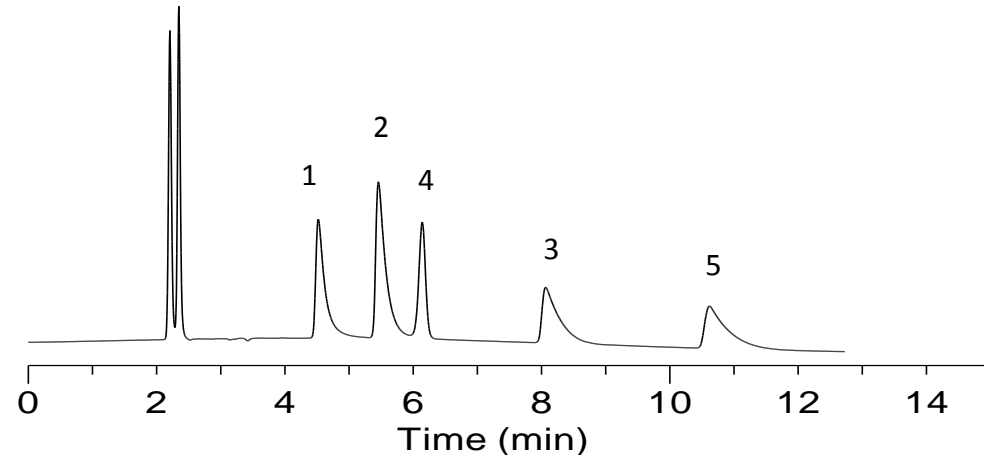
# Comparison of Strong Basic Compound Test

A basic compound test using antihistamine drugs are highly basic, which can show tailing of peaks and different elution pattern on columns with insufficient end-capping. Columns with insufficient end-capping will show later elution of sample 1, 2, 3 and 5.

## InertSustainSwift C8



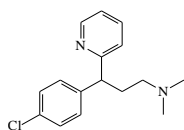
## Hypersil BDS C8



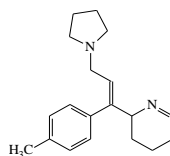
### Conditions

Eluent : A) CH<sub>3</sub>CN  
 B) 25mM K<sub>2</sub>HPO<sub>4</sub> (pH 7.0, KH<sub>2</sub>PO<sub>4</sub>)  
 A/B = 60/40, v / v

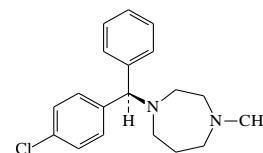
Flow Rate : 1.0 mL/min  
 Col. Temp. : 40 °C  
 Detection : UV 220 nm  
 Sample : 1. Chlorpheniramine 2. Triprolidine  
 3. Homochlorcyclizine 4. Hydroxyzine  
 5. Clemastine



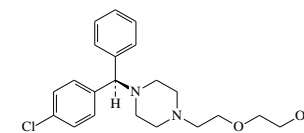
1. Chlorpheniramine



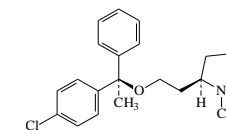
2. Triprolidine



3. Homochlorcyclizine



4. Hydroxyzine

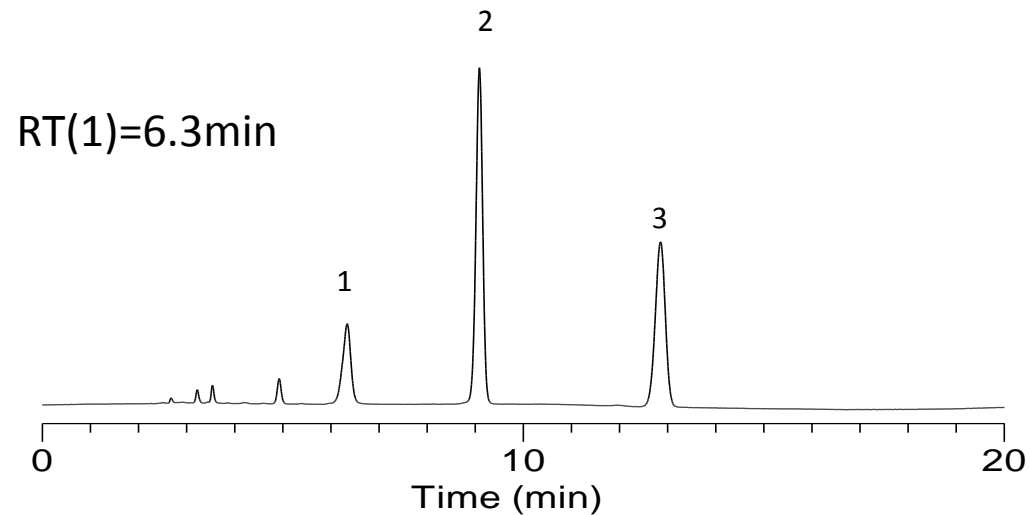


5. Clemastine

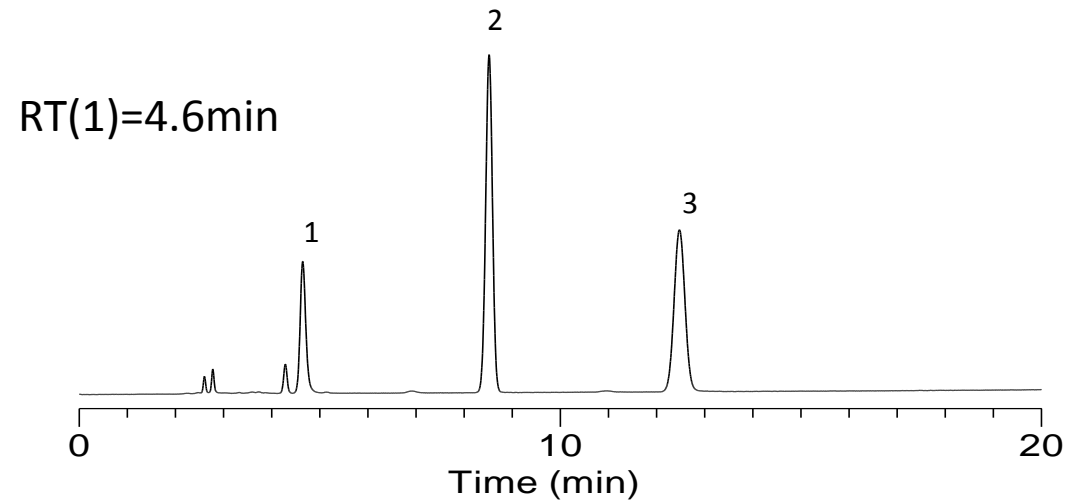
# Comparison of Strong Acidic Compound Test

Sharp peaks can be obtained when analyzing Phenol or Salicylic Acid. However, as Brilliant Blue FCF has three sulfonic groups in its chemical structure, tailing will occur when the surface of the packing material is slightly basic.

## InertSustainSwift C8

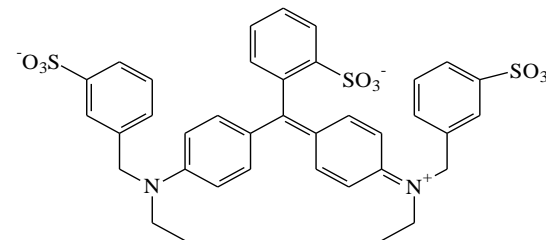


## Hypersil BDS C8



### Conditions

Eluent	: A) CH <sub>3</sub> CN B) 0.1% H <sub>3</sub> PO <sub>4</sub> A/B = 25/75, v/v	1: Brilliant Blue FCF 2: Phenol 3: Salicylic acid
Flow Rate	: 1.0 mL/min	
Col. Temp.	: 40 °C	
Detection	: UV 254 nm	

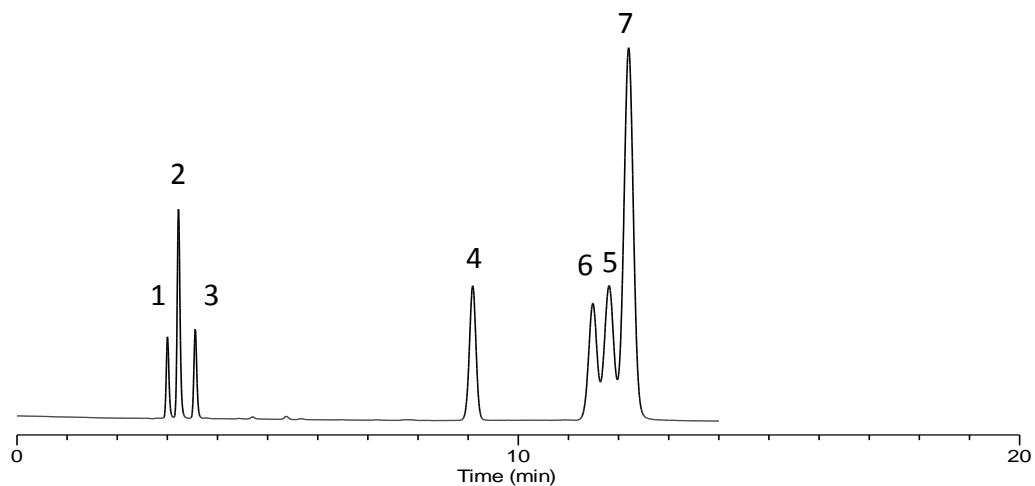


1. Brilliant Blue FCF

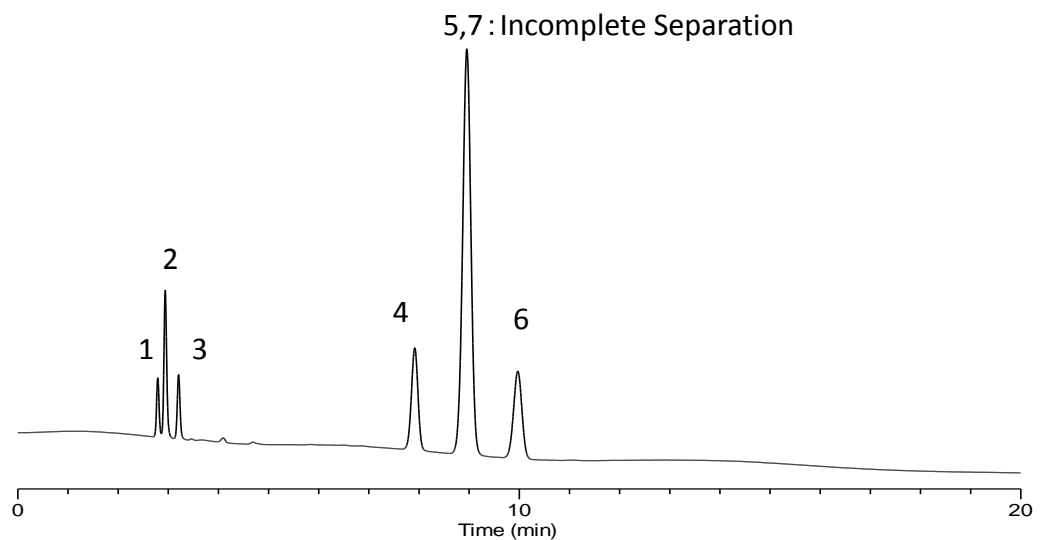
# Comparison of Selectivity

Sample No. 5, n-Amylbenzene was used to determine the hydrophobic property of columns. The overall retentivity of InertSustainSwift C8 offers similar retentivity to Hypersil BDS C8.

## InertSustainSwift C8



## Hypersil BDS C8

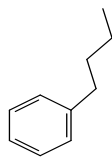


### Conditions

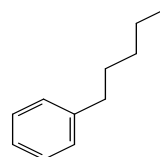
Eluent : A) CH<sub>3</sub>OH  
B) H<sub>2</sub>O  
A/B = 80/20, v/v  
Flow Rate : 1.0 mL/min  
Col. Temp. : 40 °C  
Detection : UV 254 nm

### Sample:

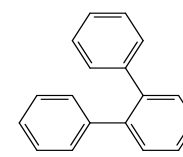
1. Uracil  
2. Caffeine  
3. Phenol  
4. n-Butylbenzene  
5. n-Amylbenzene  
6. o-Terphenyl  
7. Triphenylene



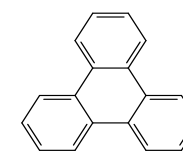
4. n-Butylbenzene



5. n-Amylbenzene



6. o-Terphenyl

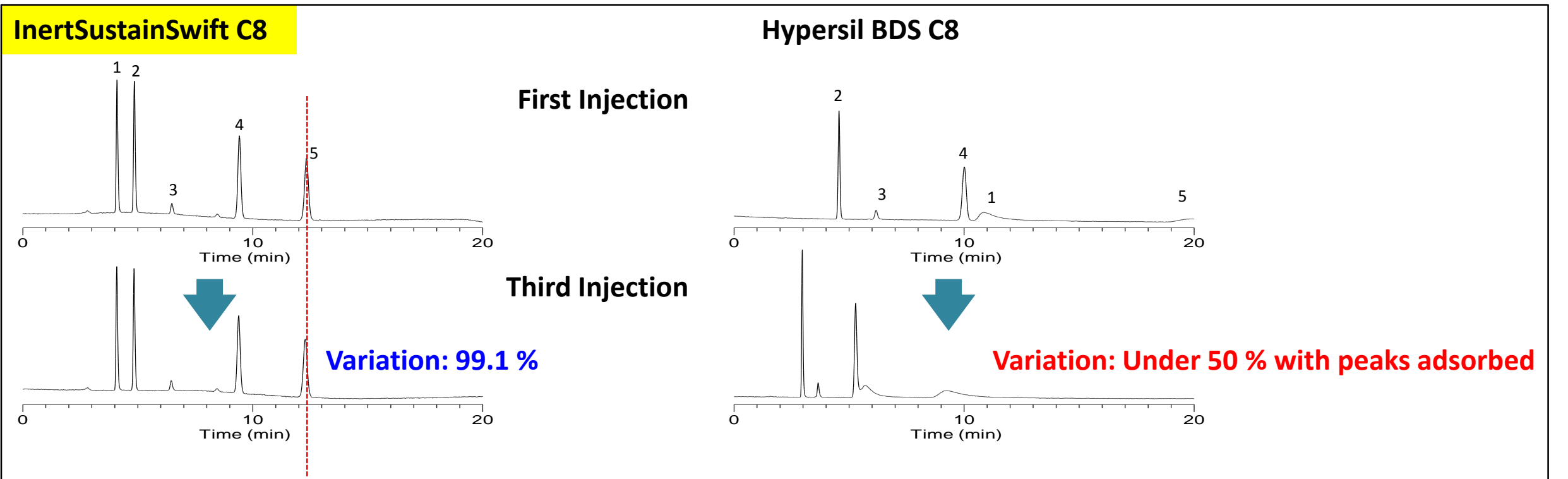


7. Triphenylene



# Comparison of Compatibility with 100 % Water Mobile Phase

Sample No. 5, n-Amylbenzene was used to determine the hydrophobic property of columns. The overall retentivity of InertSustainSwift C8 offers similar retentivity to Hypersil BDS C8.



**Conditions**

Eluent	: 100 % H <sub>2</sub> O	Sample :
Flow Rate	: 1.0 mL/min	1.Cytosine
Col. Temp.	: 40 °C	2.Uracil
Detection	: UV 254 nm	3.Guanine
		4.Thymine
		5.Adenine

- Testing Procedure**
- 1) 100 % water is introduced into column over 60 minutes.
  - 2) Conduct analysis, upper chromatogram.
  - 3) Stop flow for 15 minutes.
  - 4) 100 % water is introduced again into column over 30 minutes.
  - 5) Stop flow for 15 minutes again.
  - 6) Conduct analysis, lower chromatogram.

# Comparison of Performance

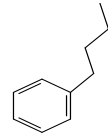
# List of Columns

Brand	Dimension (mm)	Particle Size (μm)	Surface Area (m <sup>2</sup> /g)	Pore Size (Å)	Pore Volume (mL/g)	Carbon Loading (%)	pH Range
<b>InertSustainSwfit C8</b>	<b>4.6 × 250</b>	<b>5</b>	<b>200</b>	<b>200</b>	<b>1.00</b>	<b>6</b>	<b>1-10</b>
InertSustain C8	4.6 × 250	5	350	100	0.85	8	1-10
Thermo Fisher Scientific: Hypersil BDS C8	4.6 × 250	5	170	130	0.65	7	2-9
YMC; Triart C8	4.6 × 250	5	—	120	—	17	1-12
Phenomenex; Luna C8(2)	4.6 × 250	5	400	100	—	13.5	1.5-10
YMC; YMC-Pack Pro C8	4.6 × 250	5	325	120	0.97	10	2-7.5
CERI; L-column C8	4.6 × 250	5	340	120	1.1	10	2-7.5
Waters; SunFire C8	4.6 × 250	5	340	100	0.86	11.5	2-8
Waters; XBridge C8	4.6 × 250	5	185	130	0.77	13	1-12
Agilent; Zorbax Eclipse plus C8	4.6 × 250	5	95	160	—	8	2-9

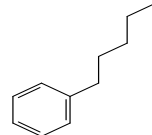
# Explanation of Analytical Tests and Conditions

## Selectivity Test

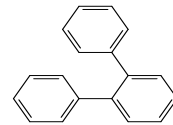
Sample No.4, n-Butylbenzene and Sample No.5, n-Amylbenzene were used to determine the hydrophobic property of the column. n-Amylbenzene elutes later against n-Butylbenzene when the hydrophobicity of the column is high. Stereoselectivity is indicated by Sample No.6, o-Terphenyl and Sample No.7, Triphenylene. O-Terphenyl has a twisted tertiary structure and Triphenylene has a planar structure. Triphenylene elutes later against o-Terphenyl when the stereoselectivity of the column is high.



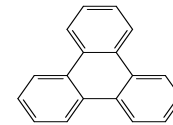
4. n-Butylbenzene



5. n-Amylbenzene



6. o-Terphenyl



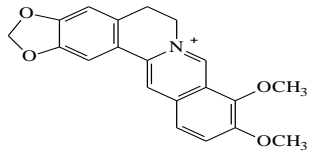
7. Triphenylene

### Conditions

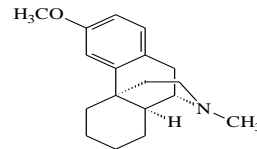
Eluent	: A) CH <sub>3</sub> OH B) H <sub>2</sub> O A/B = 80/20, v / v
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 254 nm
Sample	: 1. Uracil 2. Caffeine 3. Phenol 4. n-Butylbenzene 5. n-Amylbenzene 6. o-Terphenyl 7. Triphenylene

## Basic Compound Test (1)

Dextromethorphan and Berberine chloride are strong basic compounds. Severe tailing can be confirmed when the packing material contains residual silanol groups.



4. Berberine chloride



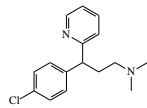
5. Dextromethorphan

### Conditions

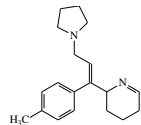
Eluent	: A) CH <sub>3</sub> CN B) 25mM K <sub>2</sub> HPO <sub>4</sub> (pH 7.0, KH <sub>2</sub> PO <sub>4</sub> ) A/B = 30/70, v/ v
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 230 nm
Sample	: 1.Uracil 2.Pyridine 3.Phenol 4.Berberine chloride 5.Dextromethorphan

## Basic Compound Test (2)

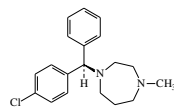
A basic compound test using antihistamine drugs and are highly basic, which can show tailing of peaks and different elution pattern on columns with insufficient end-capping. Column with insufficient end-capping will show later elution of sample 1, 2, 3 and 5.



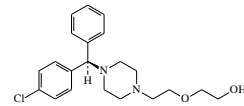
1. Chlorpheniramine



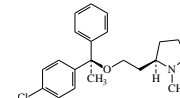
2. Triprolidine



3. Homochlorcyclizine



4. Hydroxyzine



5. Clemastine

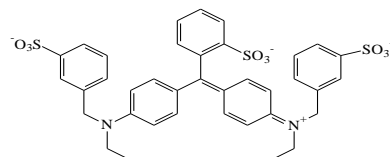
### Conditions

Eluent	: A) CH <sub>3</sub> CN B) 25mM K <sub>2</sub> HPO <sub>4</sub> (pH 7.0, KH <sub>2</sub> PO <sub>4</sub> ) A/B = 60/40, v / v
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 220 nm
Sample	: 1. Chlorpheniramine 2. Triprolidine 3. Homochlorcyclizine 4. Hydroxyzine 5. Clemastine

# Explanation of Analytical Tests and Conditions

## Acidic Compound Test

Sharp peaks can be obtained when analyzing Phenol or Salicylic Acid. However, as Brilliant Blue FCF has three sulfonic groups in its chemical structure, tailing will occur when the surface of the packing material is slightly basic.



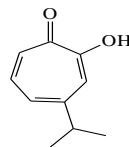
1. Brilliant Blue FCF

### Conditions

Eluent	: A) CH <sub>3</sub> CN B) 0.1% H <sub>3</sub> PO <sub>4</sub> A/B = 25/75, v/v
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 254 nm
Sample	: 1. Brilliant Blue FCF 2. Phenol 3. Salicylic acid

## Chelating Compound Test

Hinokitiol is a strong chelating compound, which coordinately binds with the surface of residual trace metal impurities, resulting in severe tailing. However, the peak shape improves as the injection increases since the surface of the packing material of the adsorption active sites eventually become masked.



2. β-Thujaaplicin (Hinokitiol)

### Conditions

Eluent	: A) CH <sub>3</sub> CN B) 0.1% H <sub>3</sub> PO <sub>4</sub> A/B = 40/60, v/v
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 254 nm
Inject Vol.	: 1 μL, 10ppm
Sample	: 1. Phenol 2. β-Thujaaplicin (Hinokitiol)

## Dewetting Test

When analyzing hydrophilic compounds under water rich mobile phase condition, once the pump is stopped, the hydrophobic bonded group pushes the aqueous mobile phase out of the pore in an irreversible fashion, in what has become known as the dewetting phenomenon.

### Testing Procedure:

- 1) 100 % water is introduced into column over 60 minutes.
- 2) Conduct Analysis (Upper chromatogram in the following pages)
- 3) Stop flow for 15 minutes.
- 4) 100 % water is introduced again into column over 30 minutes.
- 5) Stop flow for 15 minutes again.
- 6) Conduct Analysis (Lower chromatogram in the following pages)

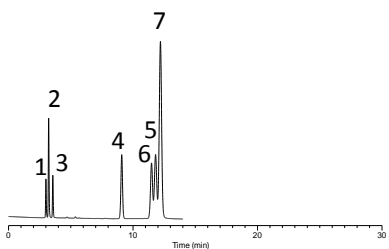
### Conditions

Eluent	: 100 % H <sub>2</sub> O
Flow Rate	: 1.0 mL/min
Col. Temp.	: 40 °C
Detection	: UV 254 nm
Sample	: 1. Cytosine 2. Uracil 3. Guanine 4. Thymine 5. Adenine

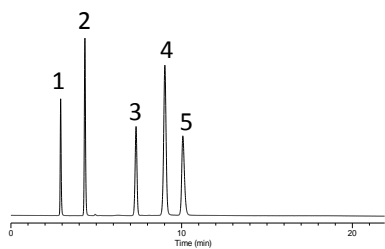
# Comparison of Performance (1/9)

## InertSustainSwift C8

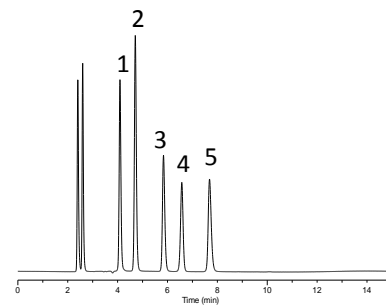
### Selectivity Test



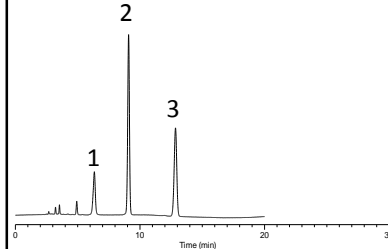
### Basic Compound Test (1)



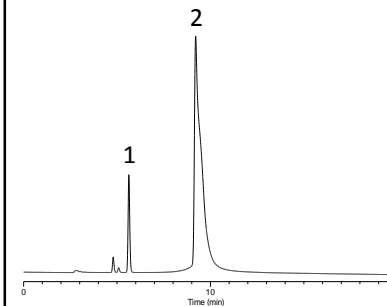
### Basic Compound Test (2)



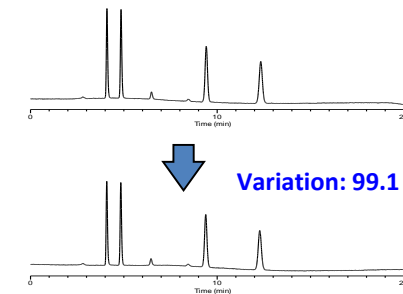
### Acidic Compound Test



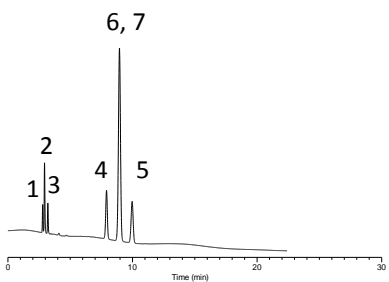
### Chelating Compound Test



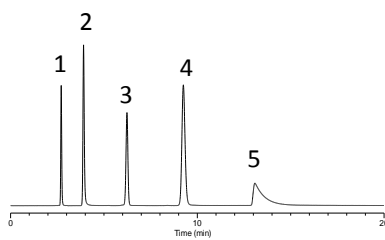
### Dewetting Test



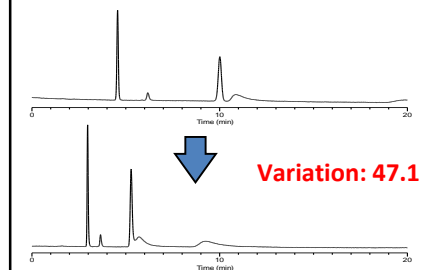
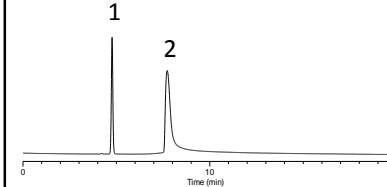
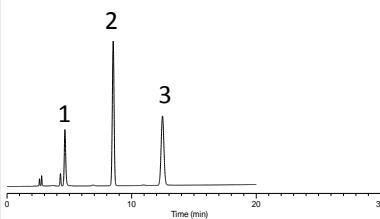
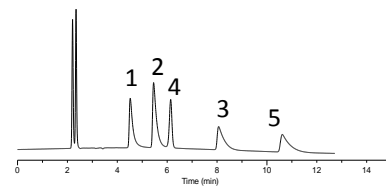
## Hypersil BDS C8



5: Adsorption observed



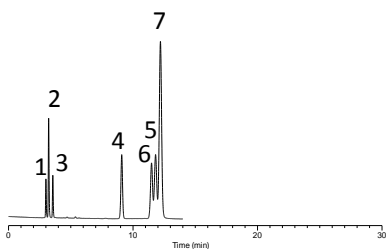
Severe tailing of peaks



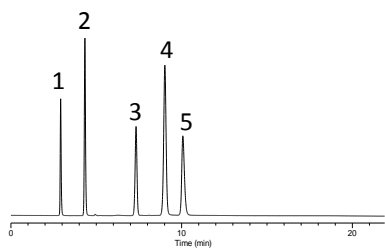
# Comparison of Performance (2/9)

## InertSustainSwift C8

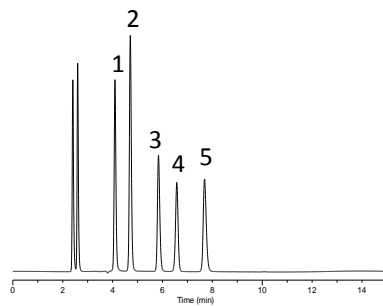
### Selectivity Test



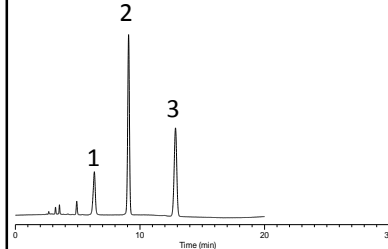
### Basic Compound Test (1)



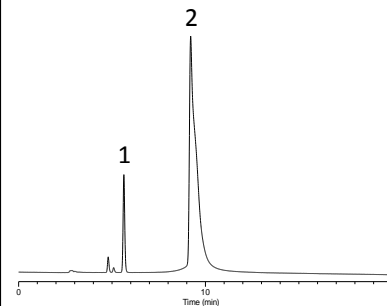
### Basic Compound Test (2)



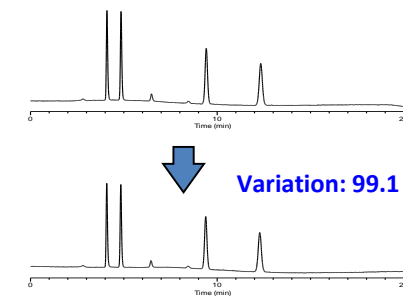
### Acidic Compound Test



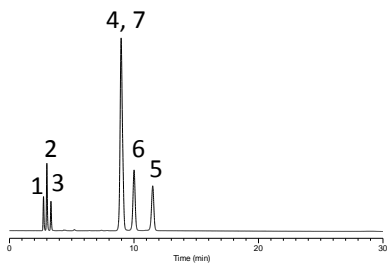
### Chelating Compound Test



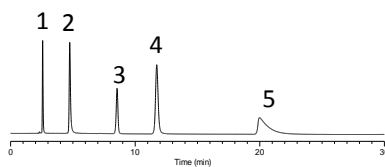
### Dewetting Test



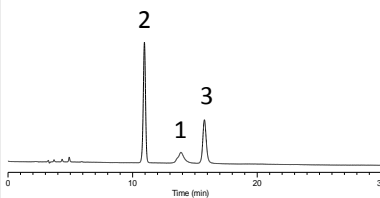
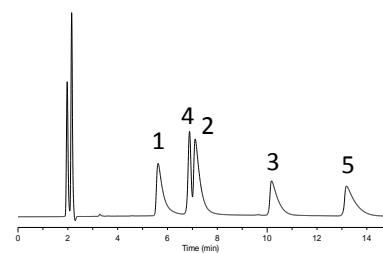
## Luna C8



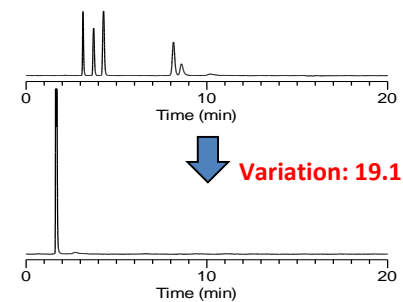
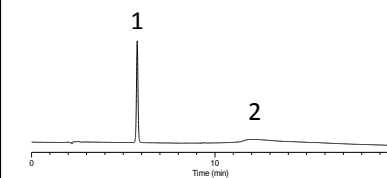
5: Adsorption observed



Severe tailing of peaks



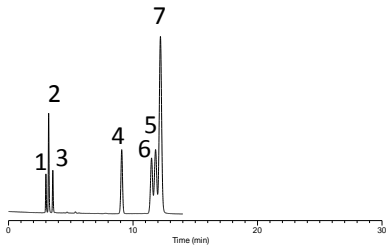
2: Adsorption observed



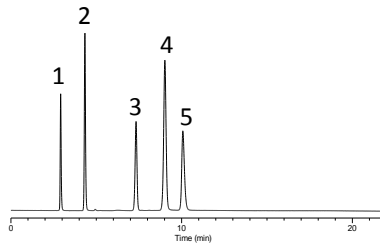
# Comparison of Performance (3/9)

## InertSustainSwift C8

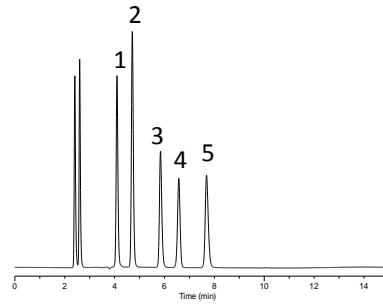
Selectivity Test



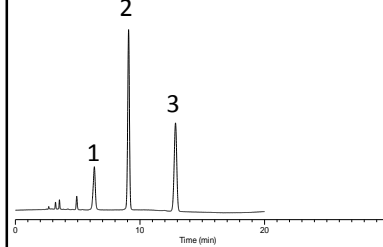
Basic Compound Test (1)



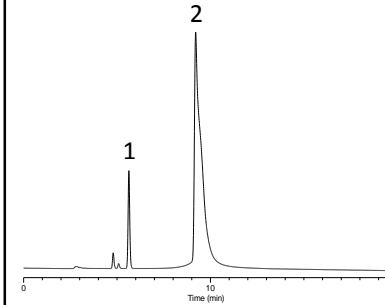
Basic Compound Test (2)



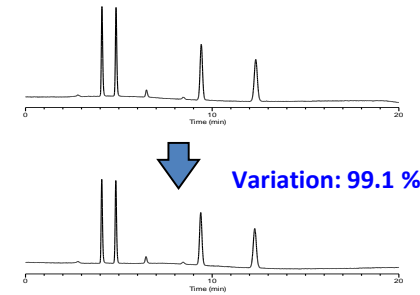
Acidic Compound Test



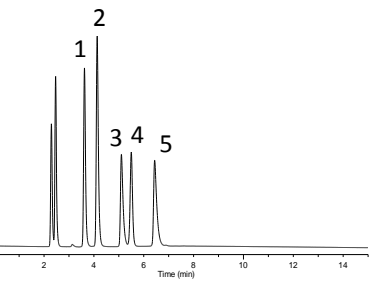
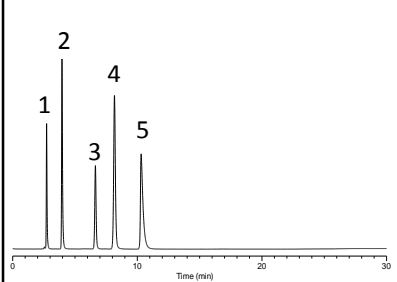
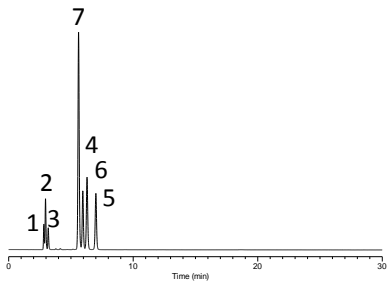
Chelating Compound Test



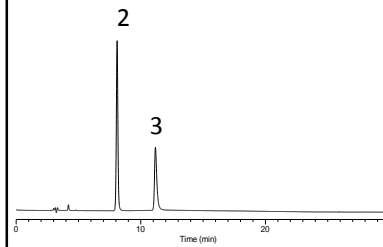
Dewetting Test



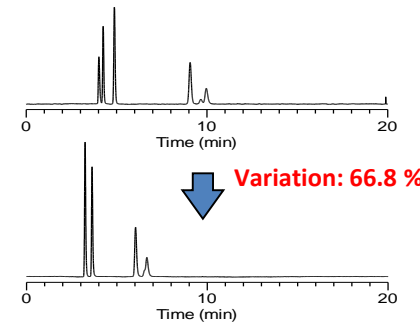
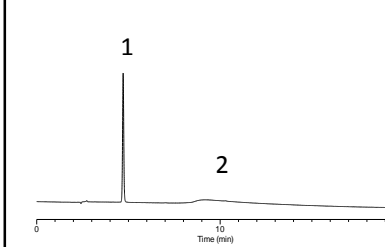
## XBridge C8



1:Not eluted



2:Adsorption observed

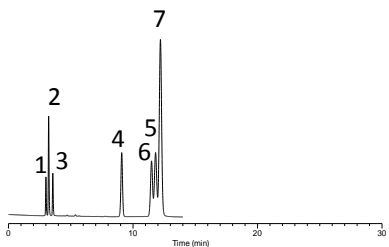




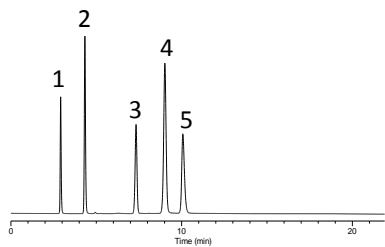
# Comparison of Performance (4/9)

## InertSustainSwift C8

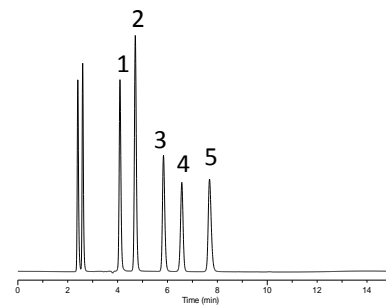
### Selectivity Test



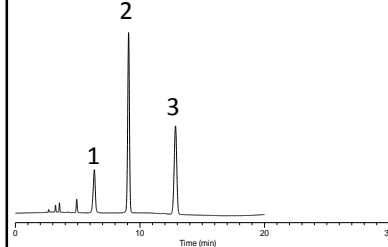
### Basic Compound Test (1)



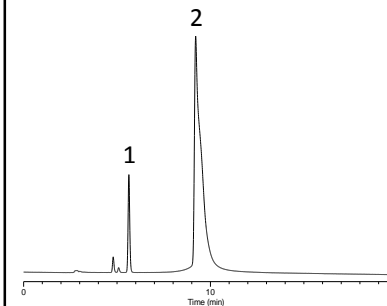
### Basic Compound Test (2)



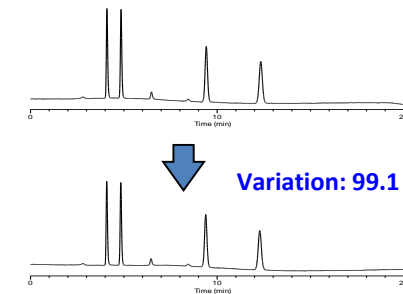
### Acidic Compound Test



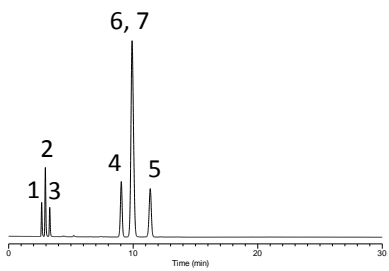
### Chelating Compound Test



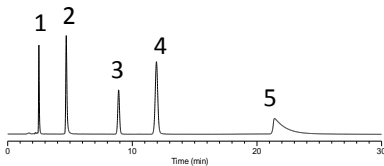
### Dewetting Test



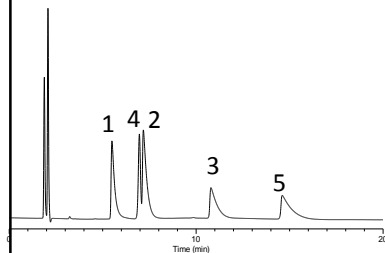
## SunFire C8



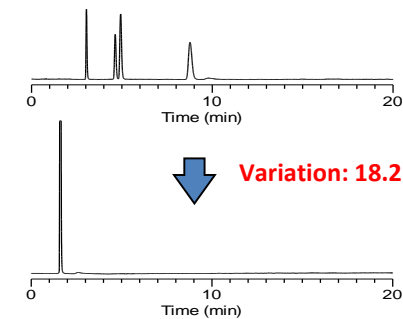
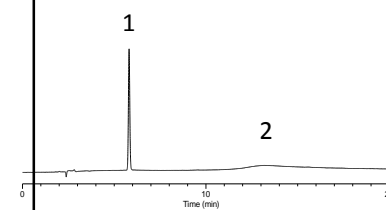
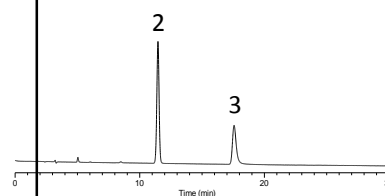
5: Adsorption observed



1: Not eluted



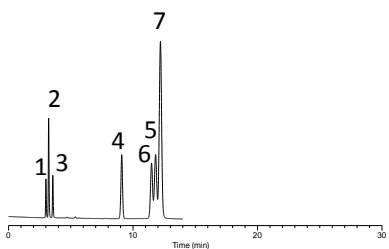
2: Adsorption observed



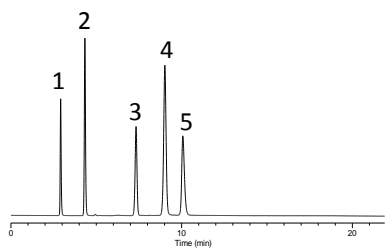
# Comparison of Performance (5/9)

## InertSustainSwift C8

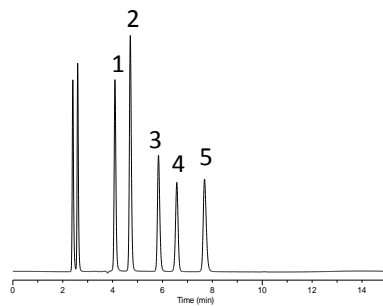
### Selectivity Test



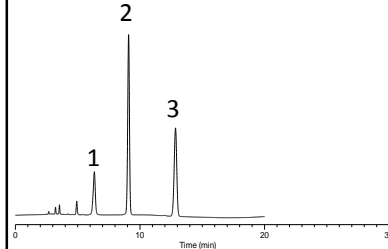
### Basic Compound Test (1)



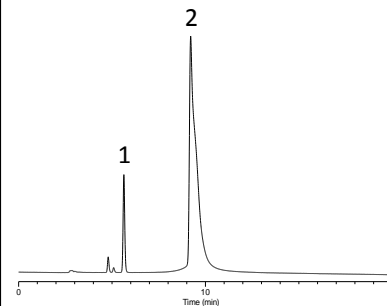
### Basic Compound Test (2)



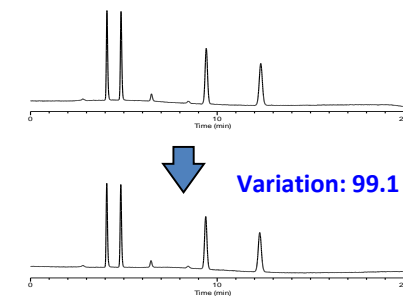
### Acidic Compound Test



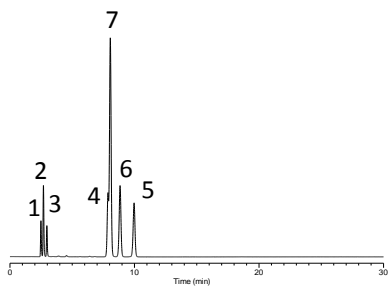
### Chelating Compound Test



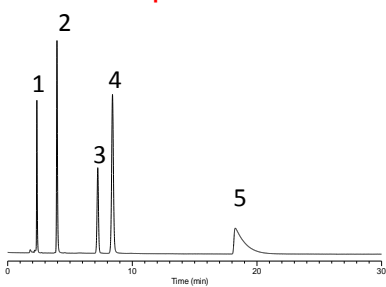
### Dewetting Test



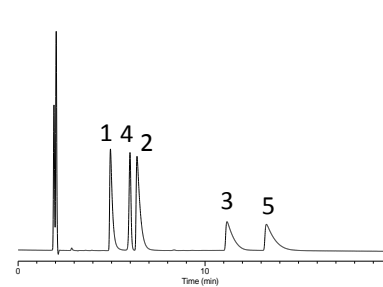
## ZORBAX Eclipse Plus C8



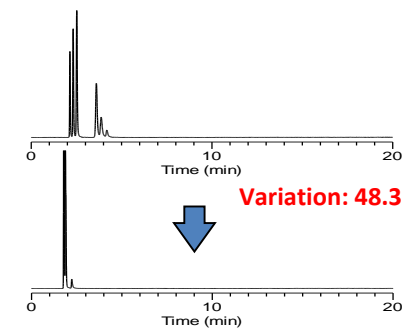
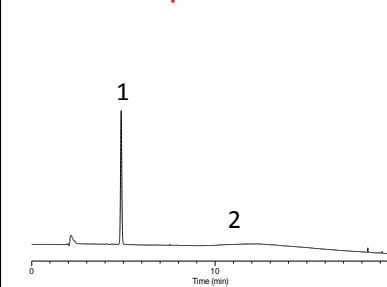
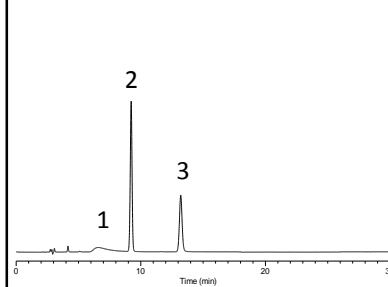
5: Adsorption observed



1: Not eluted



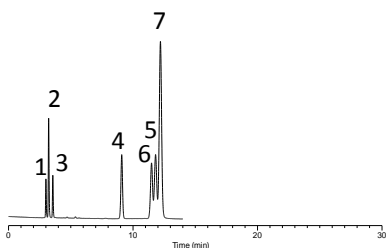
2: Adsorption observed



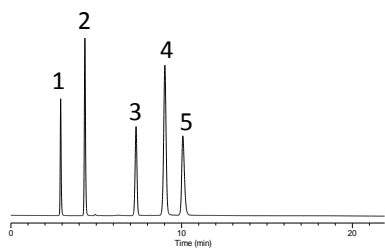
# Comparison of Performance (6/9)

## InertSustainSwift C8

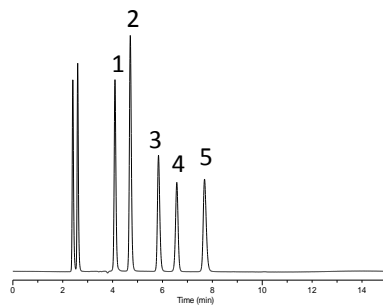
### Selectivity Test



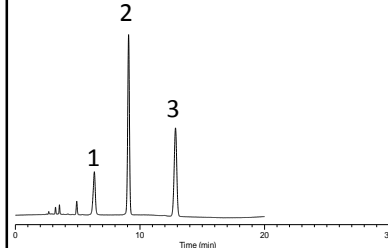
### Basic Compound Test (1)



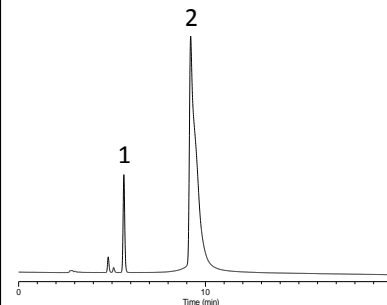
### Basic Compound Test (2)



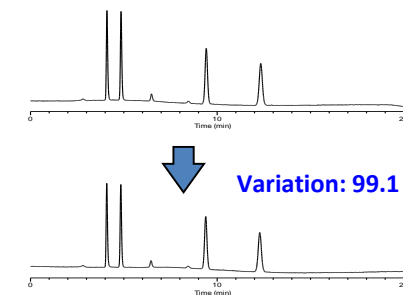
### Acidic Compound Test



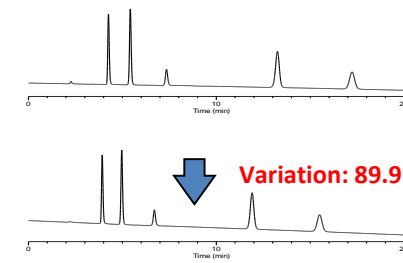
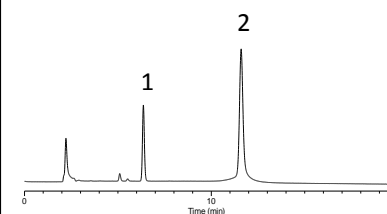
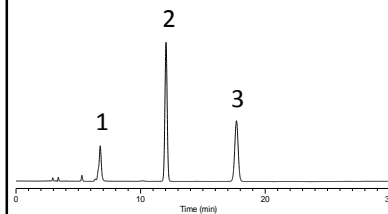
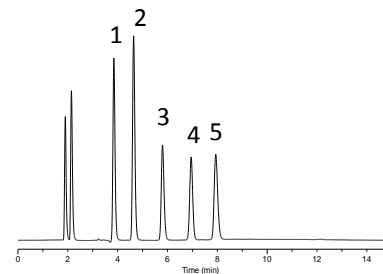
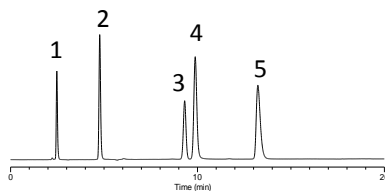
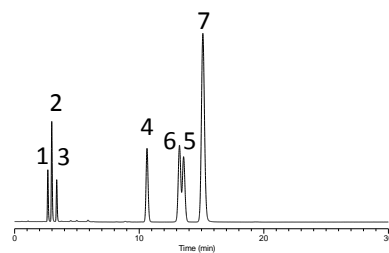
### Chelating Compound Test



### Dewetting Test

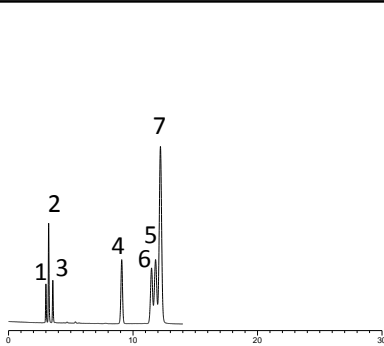
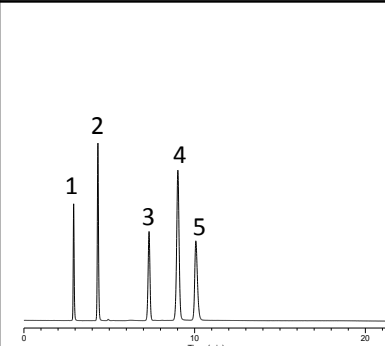
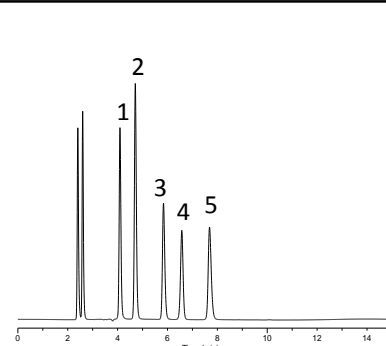
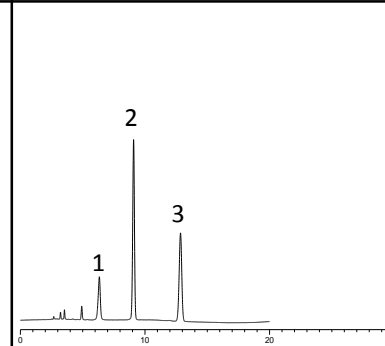
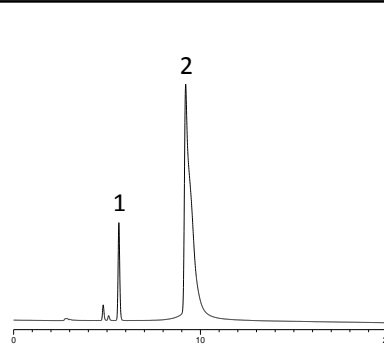
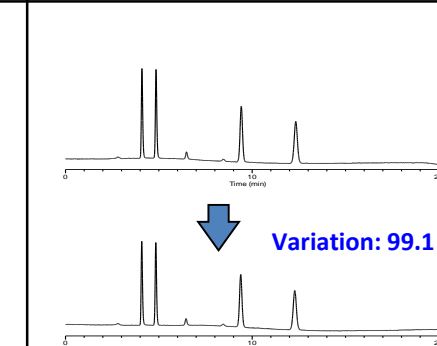
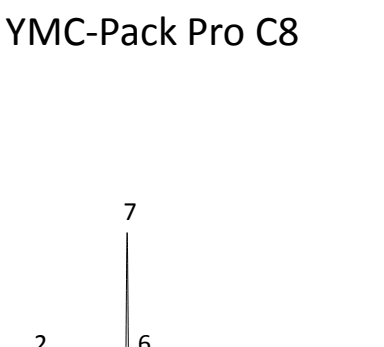
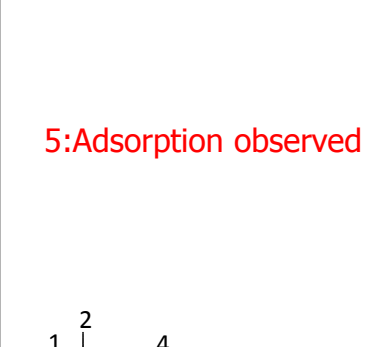
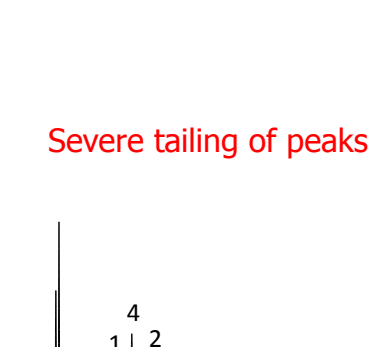
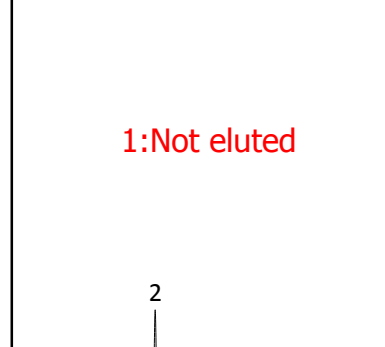
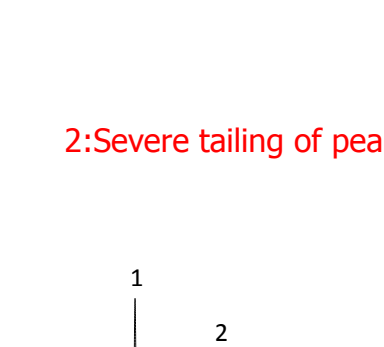
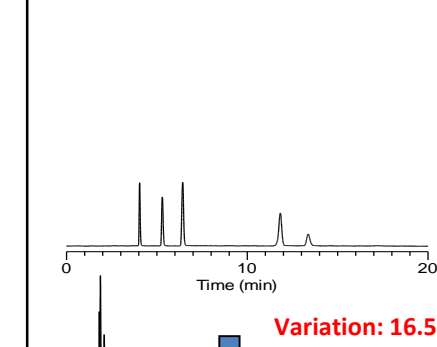


## Triart C8



# Comparison of Performance (7/9)

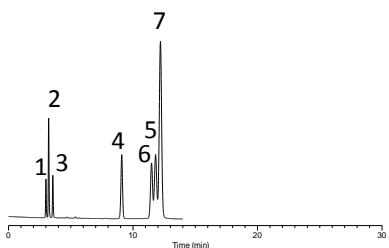
## InertSustainSwift C8

Selectivity Test	Basic Compound Test (1)	Basic Compound Test (2)	Acidic Compound Test	Chelating Compound Test	Dewetting Test
					
<h3>YMC-Pack Pro C8</h3> 	<p>5: Adsorption observed</p> 	<p>Severe tailing of peaks</p> 	<p>1: Not eluted</p> 	<p>2: Severe tailing of peak</p> 	

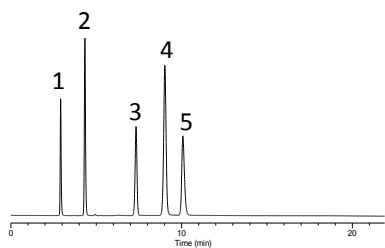
# Comparison of Performance (8/9)

## InertSustainSwift C8

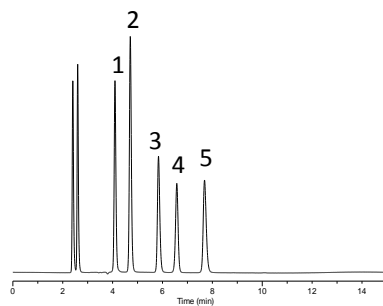
### Selectivity Test



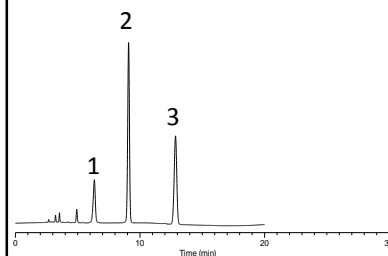
### Basic Compound Test (1)



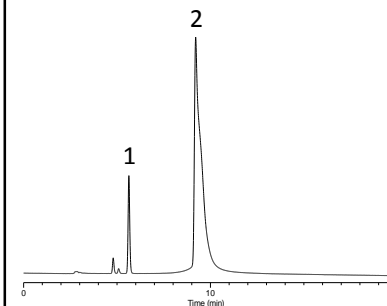
### Basic Compound Test (2)



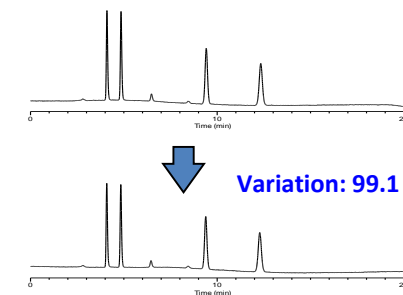
### Acidic Compound Test



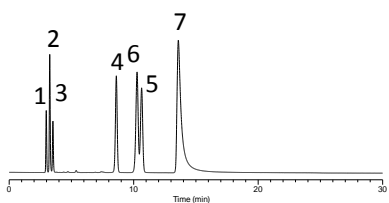
### Chelating Compound Test



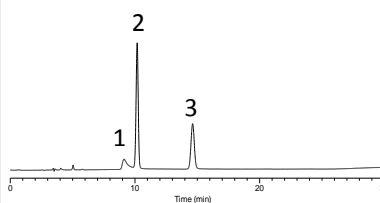
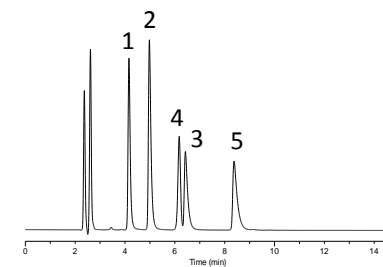
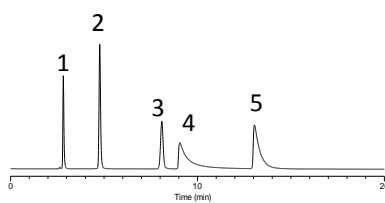
### Dewetting Test



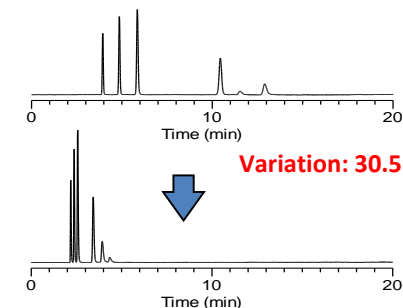
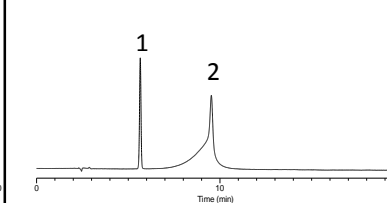
## L-column C8



4, 5: Severe tailing of peaks



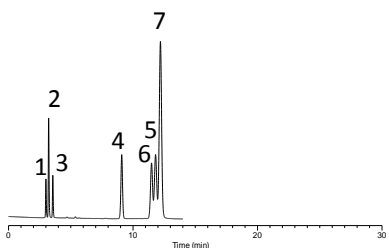
2: Severe fronting of peak



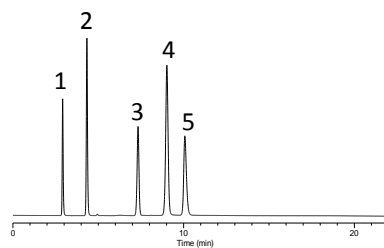
# Comparison of Performance (9/9)

## InertSustainSwift C8

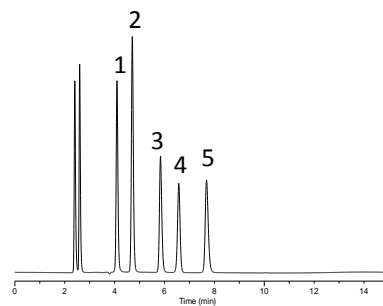
### Selectivity Test



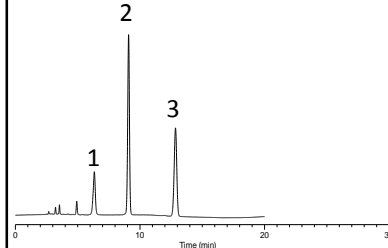
### Basic Compound Test (1)



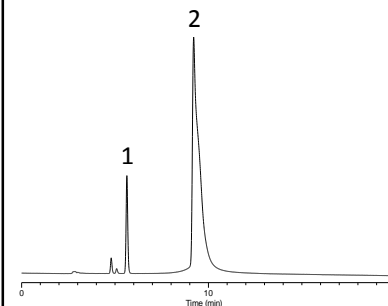
### Basic Compound Test (2)



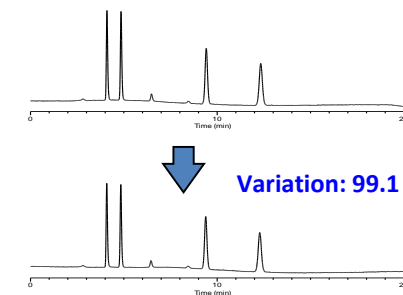
### Acidic Compound Test



### Chelating Compound Test



### Dewetting Test



## InertSustain C8

